

Download Chemistry The Periodic Law Study Guide

Start studying CHEMISTRY Chapter 5: The Periodic Law Study guide. Learn vocabulary, terms, and more with flashcards, games, and other study tools. TEACHER GUIDE AND ANSWERS Chemistry: Matter and Change Teacher Guide and Answers 7 Study Guide - Chapter 6 – The Periodic Table and Periodic Law Section 6.1 Development of the Modern Periodic Table 1. octaves 2. eight 3. nine 4. accepted 5. Dmitri Mendeleev 6. atomic mass 7. elements 8. atomic number 9. Henry Moseley 10. protons 11. periodic law Chemistry Chapter 6 Study Guide. Elements in groups 3-12 that s contained in the d-block of the periodic table and, with some exceptions, is characterized by a filled outermost s orbital of energy level n, and filled or partially filled d orbitals of energy in level n-1. Modern Chemistry 6 The Periodic Law CHAPTER 5 REVIEW The Periodic Law SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. _____ When an electron is added to a neutral atom, energy is (a) always absorbed. (c) either absorbed or released. (b) always released.